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Acids. taste sour, change the color of an acid base indicator, can be strong or weak electrolytes in aqueous solution Ex: citrus. bases. taste bitter, feel slippery, will change the color of an acid base indicator, can be strong or weak electrolytes in aqueous solution. Ex: soap

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contain hydrogen, and when placed in aqueous solution, ionize to form hydrogen ( $H^+$  ions). Chapter 19 - Acids, Bases, and Salts - 19 Assessment ...

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SECTION 19.1 ACID-BASE THEORIES (pages 587-593)  
This section compares and contrasts acids and bases as defined by the theories of Arrhenius, Brønsted-Lowry, and Lewis. It also identifies conjugate acid-base pairs in acid-base reactions.

~~Chapter 19 Acids Bases Worksheet Answers~~

Chapter 19 Acids, Bases, and Salts □□ What are the properties of acids and bases? acids taste sour, will change the color of an acid-base indicator, and can be strong or weak electrolytes in Samples

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conjugate acid: particle formed when a base gains a

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hydrogen ion: conjugate base: particle remaining when an acid donates a hydrogen ion: diprotic acid: acids containing two ionizable hydrogens: Lewis acid: substance that accepts a pair of electrons in a reaction: Lewis base: substance that donates a pair of electrons in a reaction: Arrhenius acid: substances that release hydrogen ions ( $H^+$ ) in aqueous solution: Arrhenius base

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Acid. Changes the color of an acid-base indicator (acid, base, or both) Both. Can be strong or weak electrolytes in aqueous solution. Page 10/29

~~Chapter 19 Acids Bases Salts Practice Problems~~

634 Chapter 18 □ Acids and Bases Section 118.18.1

Figure 18.1 Rhododendrons flourish in rich, moist soil that is moderately acidic, whereas semper-vivum, commonly called hen and chicks, grows best in drier, slightly basic soil. Introduction to Acids and Bases-!).)DEADifferent models help describe the behavior of acids and bases.

~~Chapter 18: Acids and Bases~~

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